## Alkaloids from the Fruits of Daphniphyllum macropodum

by Xiao-Ning Wang, Li-She Gan, Cheng-Qi Fan, Sheng Yin, and Jian-Min Yue\*

State Key Laboratory of Drug Research, Institute of Materia Medica, Shanghai Institutes for Biological Sciences, Chinese Academy of Sciences, 555 Zuchongzhi Road, Zhangjiang Hi-Tech Park, Shanghai 201203, P. R. China

(phone: +86-21-50806718; fax: +86-21-50806718; e-mail: jmyue@mail.shcnc.ac.cn)

Three new alkaloids, N-hydroxypaxdaphnine B (1), 21-O-acetylpaxdaphnine B (2), and methyl 17hydroxyhomodaphniphyllate (3), were isolated from the fresh fruits of Daphniphyllum macropodum, together with six known alkaloids. Their structures were established on the basis of extensive spectroscopic and mass-spectrometric analyses in combination with chemical transformations.

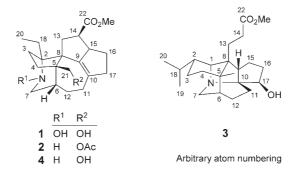
**Introduction.** – The *Daphniphyllum* alkaloids are a structurally diversified group of natural products with complex heterocyclic ring systems elaborated by plants of the genus Daphniphyllum (Daphniphyllaceae) [1]. Many of these compounds are still attracting great interest as challenging targets for total synthesis [2] as well as for biosynthetic studies [3]. In recent years, a number of novel Daphniphyllum alkaloids have been discovered from different species of this genus [4-10].

Daphniphyllum macropodum MIQ. is an evergreen shrub or small tree widely distributed in southern China [11]. The decoction of its leaves and fruits shows detoxifying, fever-clearing, and detumescent effects, and has been used in the treatment of inflammatory diseases in traditional Chinese medicine (TCM) [12]. Chemical investigations on this species growing in Japan were mainly performed from 1966 to 1975, which led to the isolation of a series of alkaloids belonging to the daphniphylline and yuzurimine types [1a]. Recently, eight additional new alkaloids have been reported from this plant collected in China [6a][8][9].

In our search for structurally unique and biogenetically interesting natural products [5], we herein report three new alkaloids, N-hydroxypaxdaphnine B (1), 21-Oacetylpaxdaphnine B (2), and methyl 17-hydroxyhomodaphniphyllate (3), from the fresh fruits of D. macropodum, together with six known alkaloids. Herein, we describe the isolation and structure elucidation of these new alkaloids.

**Results and Discussion.** – Compound **1** was isolated as a colorless amorphous solid. The molecular formula was determined as  $C_{21}H_{31}NO_4$  on the basis of HR-EI-MS (m/z 361.2264 (*M*<sup>+</sup>; calc. 361.2253)) and NMR data (*Table 1*). The IR absorption band at  $1734 \text{ cm}^{-1}$  indicated the presence of a C=O functionality. The <sup>13</sup>C-NMR data of **1** (Table 1) displayed signals for 21 C-atoms, including two Me groups, ten CH<sub>2</sub> groups (the one at  $\delta(C)$  67.4 being oxygenated, and the one at  $\delta(C)$  57.9 being linked to an Natom), three CH groups, and six quaternary C-atoms (the one at  $\delta(C)$  79.4 being linked

<sup>© 2007</sup> Verlag Helvetica Chimica Acta AG, Zürich



	1		2	
	$\delta(\mathrm{H})$	$\delta(C)$	$\delta(\mathrm{H})$	$\delta(C)$
C(2)	-	79.4	_	70.7
$CH_2(3)$	1.86 - 1.92 (m)	24.6	1.69 - 1.74 (m)	30.5
$CH_2(4)$	1.50 - 1.58 (m)	36.2	1.70 - 1.76(m)	35.6
C(5)	_	52.0	_	50.9
H-C(6)	2.13 ( <i>t</i> -like, $J = 8$ )	41.5	1.88 - 1.85 (m)	41.2
$H_a - C(7)$	3.12 (d, J = 11.0)	57.9	2.90 (d, J = 14.7)	44.9
$H_{\beta}-C(7)$	2.99 (dd, J = 11.0, 7.1)		3.16 (ddd, J = 14.7, 6.6, 1.0)	
C(8)	_	60.4	_	60.3
C(9)	_	145.2	_	145.6
C(10)	_	137.4	_	138.8
$H_{a} - C(11)$	2.50 - 2.52 (m)	27.2	2.43 - 2.48 (m)	27.3
$H_{\beta}-C(11)$	1.88 - 1.96 (m)		2.10 - 2.18(m)	
$H_{a} - C(12)$	1.47 - 1.54 (m)	28.4	1.52 - 1.56(m)	28.0
$H_{\beta}-C(12)$	2.02 - 2.05(m)		1.92 - 1.99(m)	
$H_{a}^{\prime} - C(13)$	2.03 (dd, J = 15.3, 9.4)	38.6	2.07 (dd, J = 15.2, 9.6)	37.7
$H_{\beta}-C(13)$	2.52 (dd, J = 15.3, 3.5)		2.50 (dd, J = 15.2, 3.5)	
H - C(14)	2.93 (td, J = 10.6, 3.3)	44.8	2.93 (td, J = 9.3, 3.2)	44.9
H - C(15)	3.40 - 3.50 (m)	58.9	3.36 - 3.46 (m)	58.7
$H_a - C(16)$	1.76 - 1.83 (m)	30.9	1.82 - 1.87 (m)	30.7
$H_{\beta}-C(16)$	1.15 - 1.22 (m)		1.20 - 1.32(m)	
$H_a - C(17)$	2.52 - 2.60 (m)	44.2	2.67 - 2.77 (m)	44.5
$H_{\beta}-C(17)$	2.29 (dd, J = 15.2, 8.4)		2.36 (dd, J = 15.0, 8.7)	
CH <sub>2</sub> (18)	1.66 (q, J = 7.3)	28.4	1.08 - 1.18 (m),	29.6
			1.61 - 1.69 (m)	
Me(20)	1.01 (t, J = 7.3)	12.8	0.90(t, J = 7.3)	9.8
$H_{a} - C(21)$	3.94 (d, J = 11.5)	67.4	4.48 (d, J = 11.8)	70.4
$H_{b} - C(21)$	3.82(d, J = 11.5)		4.43 (d, J = 11.8)	
C(22)	_	178.2	_	177.6
MeO	3.61(s)	52.1	3.64 (s)	52.2
AcO	_	_	2.01 (s)	21.4
			· ·	173.5

Table 1. <sup>1</sup>*H*- and <sup>13</sup>*C*-*NMR Data of* **1** and **2**. At 400/100 MHz, resp., in CD<sub>3</sub>OD;  $\delta$  in ppm, *J* in Hz. Arbitrary atom numbering.

to an N-atom, one C=O moiety at  $\delta$ (C) 178.2, and one tetrasubstituted C=C bond at  $\delta$ (C) 145.2 and 137.4).

The <sup>1</sup>H- and <sup>13</sup>C-NMR data of **1** were very similar to those of paxdaphnine B (**4**) [5a], a known compound also isolated in the present study. Compared with **4**, there was one more O-atom in the molecular formula of **1**. The structure of **1** was further determined by HMBC and ROESY experiments (*Fig. 1*), which revealed that the additional O-atom could only be attached to the N-atom. The signals of C(2) at  $\delta$ (C) 79.4 and of C(7) at  $\delta$ (C) 57.9 were both obviously downfield-shifted relative to those in **4** ( $\delta$ (C) 70.8 and 45.2, resp.) [5a], which are diagnostic of the N–OH moiety.

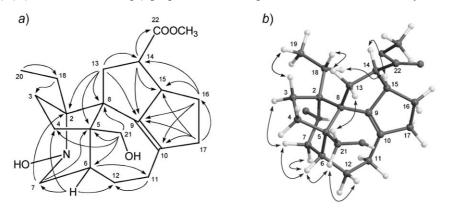


Fig. 1. Selected HMBC (a) and ROESY (b) correlations of 1

Oxidation of paxdaphnine B (4) with 3-chloroperbenzoic acid (MCPBA) afforded, indeed, compound 1 in high yield, which further confirmed its structure, including its absolute configuration [5a]. Therefore, the structure of 1 was established as N-hydroxypaxdaphnine B<sup>1</sup>).

21-O-Acetylpaxdaphnine B (2) was isolated as a colorless oil. The molecular formula  $C_{23}H_{33}NO_4$  was determined by HR-EI-MS ( $M^+$  at m/z 387.2429 (calc. 387.2410)). The <sup>1</sup>H- and <sup>13</sup>C-NMR spectroscopic data of 2 (*Table 1*) were very similar to those of 4, except for the presence of an AcO group at C(21) instead of the 21-OH group in 4. The <sup>1</sup>H- and <sup>13</sup>C-NMR signals for CH<sub>2</sub>(21) at  $\delta$ (H) 4.43 and 4.48 (2*d*, J = 11.8 Hz each), and at  $\delta$ (C) 70.4, respectively, were all severely shifted downfield relative to those in 4 ( $\delta$ (H) 3.84, 3.99 (2*d*, J = 11.6 Hz each);  $\delta$ (C) 64.7) [5a]. The signal at  $\delta$ (H) 2.01 (*s*, Me), and those at  $\delta$ (C) 21.4 and 173.5 confirmed the presence of an AcO group. The structure of 2 was further corroborated by chemical derivatization. Thus, acetylation of the parent compound 4 with acetic anhydride (Ac<sub>2</sub>O) in pyridine afforded 2 in high yield, as identified by <sup>1</sup>H-NMR comparison and co-TLC. The structure of compound 2 was, thus, assigned as 21-O-acetylpaxdaphnine B.

Methyl 17-hydroxyhomodaphniphyllate (**3**) was obtained as a yellowish oil. HR-EI-MS Analysis showed the molecular ion at m/z 375.2753, which matched the molecular formula C<sub>23</sub>H<sub>37</sub>NO<sub>3</sub> (calc. 375.2773). The IR absorption at 1736 cm<sup>-1</sup> was indicative of a C=O group. The <sup>13</sup>C-NMR (DEPT) spectrum of **3** (*Table 2*) displayed

<sup>1)</sup> For systematic names, see the *Exper. Part*.

Atom	$\delta(\mathrm{H})$	$\delta(C)$	Atom	$\delta(\mathrm{H})$	$\delta(C)$
H-C(1)	2.78–2.87 ( <i>m</i> )	64.2	$H_{\beta}-C(12)$	1.66–1.72 ( <i>m</i> )	
H-C(2)	1.27 - 1.34 (m)	38.2	$H_{a} - C(13)$	1.37 - 1.47 (m)	26.0
$H_a - C(3)$	1.40 - 1.48 (m)	27.0	$H_{b} - C(13)$	2.08 - 2.17 (m)	
$H_{\beta}-C(3)$	1.62 - 1.73 (m)		$H_{a} - C(14)$	2.46 (ddd, J = 16.5, 10.4, 4.2)	32.8
$H_a - C(4)$	1.24 - 1.33 (m)	36.6	$H_{b} - C(14)$	2.60 (ddd, J = 16.5, 10.8, 6.2)	
$H_{\beta}-C(4)$	1.79 - 1.90 (m)		$H_a - C(15)$	1.26 - 1.36(m)	22.5
C(5)	-	37.2	$H_{\beta}-C(15)$	1.48 - 1.55 (m)	
H-C(6)	1.15 - 1.23 (m)	41.6	$H_a - C(16)$	1.95 - 2.03 (m)	32.8
$H_a - C(7)$	2.76 (br. $d, J = 13.5$ )	46.8	$H_{\beta}-C(16)$	1.56 - 1.64 (m)	
$H_{\beta}-C(7)$	3.18 (br. $d, J = 13.5$ )		H - C(17)	4.08 - 4.16(m)	80.6
H-C(8)	-	48.1	H - C(18)	1.66 - 1.78 (m)	31.3
H-C(9)	2.19 - 2.28 (m)	50.8	Me(19)	1.06 (d, J = 6.6)	21.5
C(10)	-	74.6	Me(20)	0.85 (d, J = 6.6)	21.1
$H_a - C(11)$	2.07 - 2.15(m)	23.2	Me(21)	0.84(s)	25.7
$H_{\beta}-C(11)$	1.48 - 1.56 (m)		C(22)	_	174.3
$H_{a} - C(12)$	1.46 - 1.54 (m)	22.0	MeO	3.69(s)	51.5

Table 2. <sup>1</sup>*H*- and <sup>13</sup>*C*-*NMR Data of* **3**. At 400/100 MHz, resp., in C<sub>3</sub>D<sub>5</sub>N;  $\delta$  in ppm, *J* in Hz. Arbitrary atom numbering.

23 carbon signals: four Me groups (including one MeO function at  $\delta(C)$  51.5), nine CH<sub>2</sub> groups (the one at  $\delta(C)$  46.8 being linked to an N-atom), six CH groups (the one at  $\delta(C)$  80.6 being oxygenated, and the one at  $\delta(C)$  64.2 being linked to an N-atom), as well as four quaternary C-atoms (the one at  $\delta(C)$  74.6 being linked to an N-atom, as well as a C=O group at  $\delta(C)$  174.3).

The NMR spectroscopic data of **3** were similar to those reported for 17hydroxyhomodaphniphyllic acid [7b], except for an additional MeO moiety ( $\delta$ (H) 3.69 (*s*, 3 H);  $\delta$ (C) 51.5). The carbon skeleton of **3** was the same as that of 17hydroxyhomodaphniphyllic acid, as determined by analysis of the HMBC spectrum (*Fig. 2,a*). The MeO group was placed at C(22) to form a methyl ester, in accord with the observed HMBC correlation from MeO to C(22) ( $\delta$ (C) 174.3). The relative configuration of **3** was determined to be the same as that of 17-hydroxyhomodaphniphyllic acid by a ROESY experiment (*Fig. 2, b*). The structure of compound **3** was, thus, determined as methyl 17-hydroxyhomodaphniphyllate.

The following six known compounds were also isolated: paxdaphnine B (4) [5a], a zwitterionic alkaloid [13], daphnezomine S [4g], daphnilactone B [14], yuzurimine B [7b], and dehydrated daphnigracine [15]. Their structures were identified by comparison of the corresponding <sup>1</sup>H- and <sup>13</sup>C-NMR as well as MS data with those reported in the literature. Notably, compound 4, the zwitterionic alkaloid, daphnezomine S, and dehydrated daphnigracine were isolated from *D. macropodum* for the first time.

## **Experimental Part**

*General.* All solvents used were of anal. grade (*Shanghai Chemical Plant*, Shanghai, China). Thinlayer chromatography (TLC): Pre-coated silica gel  $GF_{254}$  plates (*Qingdao Haiyang Chemical Co., Ltd.*, Qingdao, China). Column chromatography (CC): silica gel (200–300 mesh), silica gel *H60*, Sephadex

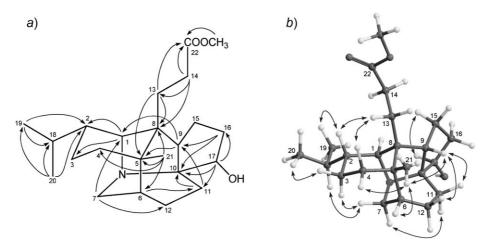


Fig. 2. Selected HMBC (a) and ROESY (b) correlations of 3

LH-20 (Amersham Biosciences), and amino silica gel (NH-DM 1020, 20–45 µm; Fuji Silysia Chemical, Ltd.). Semi-prep. HPLC: Waters system, with a 515 pump, a 2487 detector (254 nm), and an YMC-Pack ODS-A column ( $250 \times 10$  mm, S-5 µm, 12 nm). Optical rotations: Perkin-Elmer 341 polarimeter. IR Spectra: Perkin-Elmer 577 spectrometer, with KBr discs; in cm<sup>-1</sup>. NMR Spectra: Varian Mercury-Plus-400 spectrometer;  $\delta$  in ppm rel. to Me<sub>4</sub>Si, J in Hz. Electron-impact-ionization mass spectrometry (EI-MS): Finnigan MAT-95 mass spectrometer (70 eV); in m/z (rel. %). Low- and high-resolution electrospray-ionization mass spectrometers, resp.; in m/z (rel. %).

*Plant Material.* The fresh fruits of *D. macropodum* MIQ. were collected in October 2006 from Guangxi Province, P. R. China, and were authenticated by Prof. *Shao-Qing Tang* (Guangxi Normal University, P. R. China). A voucher specimen (No. DM-T-frt-zg1Y) was deposited at the Institute of Materia Medica, SIBS, Chinese Academy of Sciences.

Extraction and Isolation. The dried fresh fruits (6.0 kg) of D. macropodum were powdered and percolated with 95% EtOH to give a crude extract (320 g), which was dissolved in acidic H<sub>2</sub>O (1000 ml; adjusted to pH 4 with 0.5M H<sub>2</sub>SO<sub>4</sub>) to form a suspension. After removal of the non-alkaloidal components by extraction with AcOEt, the aq. phase was adjusted with 2M aq.  $Na_2CO_3$  soln. to pH 10, and then re-extracted with AcOEt to obtain the crude alkaloids (20.0 g). These were subjected to CC  $(SiO_2; CHCl_3/MeOH 1: 0 \rightarrow 0: 1)$  to afford five fractions (Fr. A – Fr. E). Fr. B (3.0 g) was separated by CC (SiO<sub>2</sub>; petroleum ether (PE)/AcOEt/Et<sub>2</sub>NH 10:1:0.1), and the major fraction was further purified by CC (amino silica gel; cyclohexane/CH<sub>2</sub>Cl<sub>2</sub> 3:1) to afford dehydrated daphnigracine (70 mg). Fr. C (5.0 g) was separated by CC (SiO<sub>2</sub>; PE/AcOEt/Et<sub>2</sub>NH 25:1:0.1  $\rightarrow$  4:1:0.1) to afford seven fractions (Fr. C1-Fr. C7). Fr. C2 (0.5 g) was purified by CC (Sephadex LH-20; EtOH) to afford daphnilactone B (20 mg). Fr. C3 (0.2 g) was purified by CC (amino silica gel; cyclohexane/CH<sub>2</sub>Cl<sub>2</sub> 2:1) to afford 2 (25 mg). Fr. C4 (0.9 g) was separated by CC (SiO<sub>2</sub>; PE/AcOEt/Et<sub>2</sub>NH 5:1:0.1) to afford 4 (210 mg) and 1 (20 mg). Fr. C6 (0.2 g) was repeatedly purified by CC (1. SiO<sub>2</sub>, CHCl<sub>3</sub>/MeOH 10:1; 2. Sephadex LH-20, EtOH) to provide 3 (20 mg). Fr. D (3.0 g) was recrystallized and further purified by semi-prep. HPLC (MeOH/H<sub>2</sub>O 25:75, 3 ml/min) to afford the zwitterionic alkaloid (0.7 g) and daphnezomine S (26 mg). The mother liquor was purified by CC (SiO<sub>2</sub>; CHCl<sub>3</sub>/MeOH 5:1 $\rightarrow$ 1:1) to afford yuzurimine B (35 mg).

Oxidation of Paxdaphnine B (4). 3-Chloroperbenzoic acid (MCPBA; 10 mg) was added to a stirred soln. of 4 (10 mg) in CH<sub>2</sub>Cl<sub>2</sub> (3 ml) at r.t., and the mixture was stirred for 2 h. Then, the mixture was washed with 20% aq. Na<sub>2</sub>SO<sub>3</sub> soln. (3 ml) and with H<sub>2</sub>O (6 ml). The org. layer was concentrated under reduced pressure, and the residue was purified by CC (SiO<sub>2</sub>; PE/AcOEt/Et<sub>2</sub>NH 6:1:0.1) to afford a major product (8 mg), which was identical with 1 by <sup>1</sup>H-NMR and TLC.

Acetylation of Paxdaphnine B (**4**). To a soln. of **4** (10 mg) in anh. pyridine (1 ml) was added Ac<sub>2</sub>O (0.5 ml), and the mixture was stirred at r.t. for 24 h. After evaporation of excess reagent under vacuum, the residue was separated by CC (SiO<sub>2</sub>; PE/AcOEt/Et<sub>2</sub>NH 10:1:0.1) to afford a major product (8.6 mg), which was identical with **2** by <sup>1</sup>H-NMR and TLC.

N-*Hydroxypaxdaphnine* B (= *Methyl* (2R,5S,8S,15R)-5-*Ethyl*-6-*hydroxy*-2-(*hydroxymethyl*)-6-azapentacyclo[9.5.1.0<sup>1,5</sup>.0<sup>2,8</sup>.0<sup>14,17</sup>]heptadec-11(17)-ene-15-carboxylate; **1**). Colorless, amorphous solid.  $[a]_{20}^{20} = +38.6 (c = 0.72, MeOH)$ . IR (KBr): 3423, 2949, 1734, 1624, 1437, 1383, 1198, 1171, 1039. <sup>1</sup>Hand <sup>13</sup>C-NMR: see *Table 1*. EI-MS: 361 (28, *M*<sup>+</sup>), 344 (100), 314 (64), 297 (8), 229 (32), 183 (24). HR-EI-MS: 361.2264 (*M*<sup>+</sup>, C<sub>21</sub>H<sub>31</sub>NO<sub>4</sub><sup>+</sup>; calc. 361.2253).

21-O-Acetylpaxdaphnine B (= Methyl (2R,5S,8S,15R)-2-(Acetoxymethyl)-5-ethyl-6-azapentacyclo[9.5.1.0<sup>1.5</sup>.0<sup>2.8</sup>.0<sup>14,17</sup>]heptadec-11(17)-ene-15-carboxylate; **2**). Colorless oil.  $[\alpha]_D^{20} = +49.7$  (c = 0.67, MeOH). IR (KBr): 3448w, 2933, 1736, 1439, 1365, 1246, 1169, 1034, 814. <sup>1</sup>H- and <sup>13</sup>C-NMR: see *Table 1*. EI-MS: 387 (80, *M*<sup>+</sup>), 327 (32), 314 (100), 297 (8), 229 (20), 183 (20), 149 (20). HR-EI-MS: 387.2429 (*M*<sup>+</sup>, C<sub>23</sub>H<sub>33</sub>NO<sub>4</sub><sup>+</sup>; calc. 387.2410).

*Methyl 17-Hydroxyhomodaphniphyllate* (= *Methyl (11β)-11-Hydroxydaphnan-23-oate;* **3**). Yellowish oil.  $[\alpha]_{D}^{20} = -47.5$  (c = 0.10, MeOH). IR (KBr): 3400w, 2951, 1736, 1454, 1379, 1309, 1167, 1118, 1082, 933. <sup>1</sup>H- and <sup>13</sup>C-NMR: see *Table 2*. EI-MS: 375 (64,  $M^+$ ), 360 (35), 333 (25), 302 (36), 288 (100), 274 (28), 194 (24). HR-EI-MS: 375.2753 ( $M^+$ ,  $C_{23}H_{37}NO_3^+$ ; calc. 375.2773).

Financial support from the *National Natural Science Foundation* (No. 20472093, and Key Project No. 30630072) of China is gratefully acknowledged. We thank Prof. *S.-Q. Tang* for the collection and identification of the plant material.

## REFERENCES

- a) S. Yamamura, Y. Hirata, in 'The Alkaloids', Ed. R. H. F. Manske, Academic Press, New York, 1975, Vol. 15, p. 41; b) S. Yamamura, in 'The Alkaloids', Ed. A. Brossi, Academic Press, New York, 1986, Vol. 29, p. 265; c) J. Kobayashi, H. Morita, in 'The Alkaloids: Chemistry and Biology', Ed. G. A. Cordell, Academic Press, New York, 2003, Vol. 60, p. 165.
- [2] G. A. Wallace, C. H. Heathcock, J. Org. Chem. 2001, 66, 450; C. H. Heathcock, Proc. Natl. Acad. Sci. U.S.A. 1996, 93, 1423; C. H. Heathcock, D. Joe, J. Org. Chem. 1995, 60, 1131; C. H. Heathcock, J. C. Kath, R. B. Ruggeri, J. Org. Chem. 1995, 60, 1120; C. H. Heathcock, R. B. Ruggeri, K. F. McClure, J. Org. Chem. 1992, 57, 2585; C. H. Heathcock, Angew. Chem., Int. Ed. 1992, 31, 665; C. H. Heathcock, J. A. Stanfford, D. L. Clark, J. Org. Chem. 1992, 57, 2575; D. Sole, X. Urbaneja, J. Bonjoch, Org. Lett. 2005, 7, 5461.
- [3] H. Niwa, Y. Hirata, K. T. Suzuki, S. Yamamura, *Tetrahedron Lett.* 1973, 14, 2129; K. T. Suzuki, S. Okuda, H. Niwa, M. Toda, Y. Hirata, S. Yamamura, *Tetrahedron Lett.* 1973, 14, 799.
- [4] a) S. Saito, T. Kubota, E. Fukushi, J. Kawabata, H. Zhang, J. Kobayashi, Org. Lett. 2007, 9, 1207; b) S. Saito, T. Kubota, J. Kobayashi, Tetrahedron Lett. 2007, 48, 38092; c) S. Saito, T. Kubota, E. Fukushi, J. Kawabata, H. Zhang, J. Kobayashi, Tetrahedron Lett. 2007, 48, 1587; d) H. Morita, N. Ishioka, H. Takatsu, T. Iizuka, J. Kobayashi, J. Nat. Prod. 2006, 69, 418; e) T. Kubota, Y. Matsuno, H. Morita, T. Shinzato, M. Sekiguchi, J. Kobayashi, Tetrahedron 2006, 62, 4743; f) H. Morita, N. Ishioka, H. Takatsu, T. Shinzato, Y. Obara, N. Nakahata, J. Kobayashi, Org. Lett. 2005, 7, 459; g) H. Morita, H. Takatsu, J. Kobayashi, Tetrahedron 2003, 59, 3575.
- [5] a) C. Q. Fan, S. Yin, J. J. Xue, J. M. Yue, *Tetrahedron* 2007, 63, 115; b) H. Zhang, S. P. Yang, C. Q. Fan, J. Ding, J. M. Yue, *J. Nat. Prod.* 2006, 69, 553; c) S. P. Yang, H. Zhang, C. R. Zhang, H. D. Cheng, J. M. Yue, *J. Nat. Prod.* 2006, 69, 79; d) Z. J. Zhan, C. R. Zhang, J. M. Yue, *Tetrahedron* 2005, 61, 11038; e) X. Chen, Z. J. Zhan, J. M. Yue, *Helv. Chim. Acta* 2005, 88, 854; f) X. Chen, Z. J. Zhan, J. M. Yue, *Helv. Chim. Acta* 2005, 88, 854; f) X. Chen, Z. J. Zhan, J. M. Yue, *Chem. Biodiv.* 2004, 1, 1513; g) Z. J. Zhan, S. P. Yang, J. M. Yue, *J. Org. Chem.* 2004, 69, 1726; h) S. P. Yang, J. M. Yue, *Org. Lett.* 2004, 6, 1401; i) S. P. Yang, J. M. Yue, *J. Org. Chem.* 2003, 68, 7961; j) S. P. Yang, J. M. Yue, *Helv. Chim. Acta* 2006, 89, 2783.
- [6] a) N. C. Kong, H. P. He, Y. H. Wang, S. Gao, Y. T. Di, X. J. Hao, *Helv. Chim. Acta* 2007, *90*, 972;
  b) S. Z. Mu, X. W. Yang, Y. T. Di, H. P. He, Y. Wang, Y. H. Wang, L. Li, X. J. Hao, *Chem. Biodiv.*

**2007**, *4*, 129; c) C. S. Li, H. P. He, Y. T. Di, Y. H. Wang, S. Z. Mu, S. L. Li, S. Gao, Z. L. Gao, X. J. Hao, *Tetrahedron Lett.* **2007**, *48*, 2737; d) Y. T. Di, H. P. He, Y. S. Wang, L. B. Li, Y. Lu, J. B. Gong, X. Fang, N. C. Kong, S. L. Li, H. J. Zhu, X. J. Hao, *Org. Lett.* **2007**, *9*, 1355; e) L. Li, H. P. He, Y. T. Di, J. M. Tian, X. J. Hao, *Helv. Chim. Acta* **2006**, *89*, 1457.

- [7] a) H. E. Bitar, V. H. Nguyen, A. Gramain, T. Sévenet, B. Bodo, *Tetrahedron Lett.* 2004, 45, 515;
   b) H. E. Bitar, V. H. Nguyen, A. Gramain, T. Sévenet, B. Bodo, *J. Nat. Prod.* 2004, 67, 1094.
- [8] Z. Y. Li, P. Chen, H. G. Xu, Y. M. Yang, S. Y. Peng, Z. Z. Zhao, Y. W. Guo, Org. Lett. 2007, 9, 477; W. Zhang, Y. W. Guo, K. Krohn, Chem.-Eur. J. 2006, 12, 5122.
- [9] X. W. Gan, H. Y. Bai, Q. G. Chen, L. Ma, L. H. Hu, Chem. Biodiv. 2006, 3, 1255.
- [10] A. Jossang, H. E. Bitar, V. C. Pham, T. Sévenet, J. Org. Chem. 2003, 68, 300.
- [11] M. Zheng, T. L. Min, in 'Flora of China' [Zhongguo Zhiwu Zhi], Science Press, Beijing, 1980, Vol. 45(1), p. 1.
- [12] The Editorial Committee of the Administration Bureau of Traditional Chinese Medicine, 'Chinese Materia Medica' [Zhonghua Benchao], Shanghai Science and Technology Press, Shanghai, 1998, Vol. 4, p. 866.
- [13] S. Yamamura, M. Toda, Y. Hirata, Bull. Chem. Soc. Jpn. 1976, 49, 839.
- [14] H. Morita, N. Yoshida, J. Kobayashi, Tetrahedron 2000, 56, 2641.
- [15] S. Yamamura, J. A. Lamberton, H. Irikawa, Bull. Chem. Soc. Jpn. 1977, 50, 1836.

Received June 20, 2007